# **Integrating Safety and Health Indexes in Computer-Aided Molecular Design Under Property Prediction Uncertainty**

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#### ABSTRACT

Computer-Aided Molecular Design (CAMD) is a powerful tool for designing promising molecules with the targeted properties. Besides properties, the safety and health aspects must also be considered during design stage to ensure the synthesized molecules do not cause much harm. The molecules are assessed by several safety and health parameters, which are measured using physicochemical properties that can be estimated via property prediction methods. Each parameter/sub-index is allocated with a penalty score based on the property to represent the hazard level of molecules. However, uncertainty in the prediction models can adversely affect the accuracy of scores. The main highlight of this paper is to manage the effect of uncertainty on the safety and health sub-indexes, by improving the scores at the uncertain region for enhanced measurement of molecular hazards. A case study on solvent design for carotenoids extraction from Palm Pressed Fiber (PPF) has been considered with the proposed methodology.

Keywords: Carotenoids Extraction, Computer-Aided Molecular Design, Property Prediction Model, Safety and Health Index, Uncertainty.

## **1.0 INTRODUCTION**

Many industrial disasters in the past were caused by the mishandling of hazardous substances, combustible dusts and reactive chemicals.1 The occurrences of such accidents have emphasized the importance of process safety as contributing factor in process design and development. The increase in complexity of processing plants due to continuing technological and social development has also resulted in new hazards and increased risk in chemical processing plants.<sup>2</sup> As a result, hazard identification and analysis methods are introduced to manage hazards in the process. These techniques may be able to reduce the overall safety impacts of the plant, but it does not completely eliminate hazards present in the plant. Since many accidents are linked to the use of harmful chemicals, one way to address this is to replace the highly hazardous chemicals with less harmful ones. The substitution of dangerous chemicals can be accomplished by utilizing chemical product design techniques. The traditional product design technique involves trial-and-error laboratory synthesis and test methodology to synthesize a set of promising molecular candidates that attain the targeted results. However, this technique is unable to excel in the present global business environment as the need of experimental works is timeconsuming and costly.3 An alternate way is the top-down reverse engineering approach, which couples with computer-aided molecular design (CAMD) technique to identify the optimal molecular structures.4 CAMD aims to determine a molecule that meets the specified target properties from a given set of molecule

building blocks.<sup>5</sup> CAMD tools utilize optimization methods to generate molecules and applied property prediction models to evaluate the molecular performance based on the set of priori target property values.<sup>6</sup> In the past few decades, the aspects of environment, safety and health have received much increased attention in many CAMD problems. Duvedi and Achenie<sup>7</sup> applied CAMD technique to design environmentally-friendly refrigerants. As for safety and health, Karunanithi *et al.*,<sup>8</sup> included flammability and toxicity as design factors to synthesize solvents for solution crystallization.

Recently, Ten et al.,9 considered the concept of inherent safety and occupational health as design objectives to evaluate the safety and health performance of the generated molecules. The aim is to ensure that the synthesized molecules have reduced hazards and do not cause much safety and health risks. The concept of Inherent Safety Design (ISD) is to eliminate or minimize hazards present in process plant by substituting hazardous substances with less dangerous materials. Numerous inherent safety indexes have been established to identify and quantify the hazard level in a process route. The pioneer of all inherent safety indexes was the Prototype Index for Inherent Safety (PIIS) developed by Edwards and Lawrence,<sup>10</sup> which quantifies and ranks the inherent safety of several process routes during conceptual design stage. The Inherent Safety Index (ISI) by Heikkilä<sup>11</sup> considered a wider scope of safety parameters that are readily available during preliminary design stage. As for occupational health, International Labor Organization (ILO) has predicted that over two million people around the world

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die yearly due to work-related diseases. Therefore, inherent occupational health has been introduced to minimize the health hazards posed by chemical processes to the workers. The Process Route Healthiness Index (PRHI) by Hassim and Edwards<sup>12</sup> was developed to assess possible occupational health hazards arisen from process routes in early design phase. Meanwhile, the Inherent Occupation Health Index (IOHI) by Hassim and Hurme<sup>13</sup> evaluates the possible health risks of different process routes during research and development stage. Similarly to inherent safety indexes, both PRHI and IOHI used health based parameters to conduct the occupational health assessment of a process route.

In the inherent indexes, the chemical-related safety and health parameters are applied to measure the molecular performance. Each parameter/sub-index is assessed using one or few molecular properties, and subjective scaling is used where the properties are divided into a few ranges. Each range is then allocated with a score depending on the degree of potential hazards. Since all properties are estimated through property prediction methods, the accuracy of the estimated property values may not always be guaranteed. Generally, there are 5-10% (or higher) of discrepancies between the actual experimental values and property predictions.<sup>14</sup> Since the properties affect the sub-index scores, thus the allocation of scores is dependent on the accuracy of the predicted values. The deviation of predicted value from its actual experimental value may cause the assignment of inaccurate score to the molecule. As a result, the main objective of this paper is to manage uncertainty resulted from property prediction in the safety and health sub-indexes. The goal is to improve the scorings for better representation of molecular hazard level.

# 2.0 METHODOLOGY

#### 2.1 Problem Formulation

The needs of a chemical product are first specified by defining the product specifications to determine the functionality and physical behavior of a product. These product specifications can be translated in terms of target properties, which are represented by the physicochemical properties of the molecules. Target properties are usually represented in terms of property range, in which the property value must fall within the predefined range for the molecule to function as intended. The target properties are then categorized as objective functions to be optimized or property constraints to be fulfilled. The possible molecular groups acting as the potential building blocks will be selected. Structural constraints are also introduced to eliminate combination of infeasible solution.

## 2.2 Selection of Inherent Safety and Health Sub-Indexes

For the selection of safety and health sub-indexes, the properties involved in the assessment of molecules must be able to be estimated through property prediction models. The safety and health sub-indexes chosen are shown in Figure 1. The properties used for the measurement of the sub-indexes are also listed in the figure. The total penalty score of a molecule (*Ishi*) is the summation of all seven sub-index scores assigned to it. A molecule with lower total penalty score is desired as it indicates a molecule with inherently safer and healthier properties.

# 2.3 Property Prediction Models and Uncertainty Management

All target properties in Section 2.1 and properties assessed in Figure 1 have to be calculated through the property prediction methods. One conventionally used approach is the group contribution method (GCM), which is able to estimate the physicochemical properties of a molecule based on its molecular structure.<sup>15</sup> The general equation for the applied GCM is shown in Equation (1). f(p) denotes the simple function of property p, while  $C_i$  represents the contribution of molecular group of type-i with  $N_i$  occurrence. Flash point  $(F_p)$ , boiling point  $(T_b)$  and melting point  $(T_m)$  are estimated using models developed by Hukkerikar *et al.*,<sup>16</sup> Meanwhile, permissible exposure limit



Figure 1: The selected safety and health sub-indexes and the properties used for assessment

(*PEL*) and  $LD_{50}$  for acute oral toxicity can be estimated using models presented by Hukkerikar *et al.*,<sup>17</sup> Conte *et al.*,<sup>18</sup> have established a model to calculate viscosity ( $\eta$ ). Both upper and lower exposure limits (*UEL* and *LEL*) can be estimated using models proposed by Frutiger *et al.*,<sup>19</sup>

$$f(p) = \sum_{i} N_i C_i \tag{1}$$

For all these GCM models, the estimated value may not always be close to the actual value. The deviation of the estimated value may result in an inaccurate sub-index value being allocated to the molecule. For instance, given that molecule K has an estimated  $\eta$  value of 7.7 cP. As shown in Figure 2(a), its  $I_n$  sub-index value would be two. However, the GCM for  $\eta$  has a standard deviation of 3.44 cP.<sup>18</sup> The exact  $\eta$ value for molecule K can fall within the range of 7.7 cP  $\pm$  3.44 cP, which is represented by the grey region in Figure 2(a). Based on Figure 2(a), it is possible that its  $I_{\eta}$  sub-index value can either be two or three, since 10 cP is a boundary value where any  $\eta$ values higher than it are allocated an  $I_{\eta}$  value of three, whereas any  $\eta$  values lower than it (but above 1 cP) are assigned an  $I_{\eta}$ value of two. Thus from the diagram, when the property value moves across the boundary value, there is an abrupt change on the sub-index value. Therefore, the allocation of sub-index value to any property value that is near to the boundary value is highly uncertain. Hence, the management of property prediction uncertainty is carried out on this region. First, the standard deviation of GCM for  $\eta$  has to be identified, which is 3.44 cP. This value is added to or subtracted from the boundary value, 10 cP, to create the uncertain range (6.56 cP to 13.44 cP, the grey zone in Figure 2(b)). When  $\eta$  is 6.56 cP, the initial  $I_{\eta}$  value is two; when  $\eta$  is 13.44 cP, the initial  $I_{\eta}$  value would be three. In the uncertain range, the  $I_{\eta}$  value linearly transits from two (at 6.56 cP) to three (at 13.44 cP). This transition is represented by the slope drawn on the uncertain range.

Meanwhile, Figure 3(a) illustrates the scorings of  $I_{AH}$  subindex. The standard deviation of GCM for log LD<sub>50</sub> is 0.43.<sup>17</sup> The management of uncertainty is done on the two boundary values (2.7 and 3.3) as shown in Figure 3(a), and the uncertain range for both boundary values are determined. However, there exists an overlapping of uncertain range for both boundary values (darker grey region in Figure 3(a)). Two linear slopes are drawn on the uncertain range for both boundary values. As shown in Figure 3(b), the overlapping region contains two different transition slopes contributed by each boundary value. However, any property value at any region can only receive a single subindex value to indicate its hazard level. Hence, composite curve is applied on this overlapping region, which is shown in Figure 3(c). The management of uncertainty is repeated for all boundary values of all sub-indexes but IFL and IEX sub-indexes, as both subindexes are each measured by two properties. Each GCM model has its distinct statistical performance indicator values and to address different uncertainties originated from multiple GCM models, the CAMD model would become complex.

# 2.4 Optimization Model

In this paper, the CAMD problem considers multiple design



Figure 2: (a) Initial form of  $I_{\eta}$  sub-index; (b)  $I_{\eta}$  sub-index with the incorporation of uncertainty



Figure 3: (a) Initial form of IAH sub-index; (b) IAH sub-index with the incorporation of uncertainty; (c) IAH sub-index with composite curve

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objectives, in which the product functionality and its safety and health attributes are optimized simultaneously. A decision making has to be made on the trade-off between different objectives. In this case, fuzzy optimization is applied to solve this CAMD problem with conflicting objectives. Each design objective is expressed in terms of a target property. First, a degree of satisfaction,  $\lambda_p$  is introduced to each design objective, which can be expressed as a linear membership function bounded by the lower and upper bounds of its corresponding target property as shown in Equations (2) and (3). Equation (2) is applied on target properties to be minimized while Equation (3) is used on target properties to be maximized.

$$\frac{V^{U} - V_{p}}{V^{U} - V^{L}} \ge \lambda_{p}$$
<sup>(2)</sup>

$$\frac{V_p - V^{\perp}}{V^{\cup} - V^{\perp}} \ge \lambda_p \tag{3}$$

Note that p represents target property p,  $V_p$  represents target property value, and  $V^{L}$  and  $V^{U}$  represent the lower and upper bounds of the target property respectively. Max-min aggregation method presented by Zimmermann<sup>20</sup> is then applied to maximize all  $\lambda_p$ . This can be done by maximizing the least satisfied degree of satisfaction, so that all  $\lambda_p$  are satisfied partially to at least the degree of  $\lambda$ . Hence, the overall objective now is to maximize the least satisfied objective, which is equivalent to  $\lambda$ .

## 3.0 CASE STUDY

## 3.1 Optimization Formulation

The objective of this case study is to identify a solvent that can replace hexane to extract carotenoids from the residual oil found in Palm Pressed Fiber (PPF). Currently, hexane is the conventionally used solvent for extraction as it offers high carotenoids extraction yield and has convenient  $T_b$  (relatively low to limit heat consumption during solvent recovery but high enough to prevent much solvent losses during extraction).<sup>21</sup> However, hexane is toxic to aquatic life, highly flammable and may result in fatality if swallowed or enters airways. The solubility of carotenoids in the solvent can be determined by the Hansen Solubility Parameters (HSP). The three parameters in HSP are  $\delta_d$ ,  $\delta_p$  and  $\delta_h$ , which represent dispersion, polar and hydrogen bonding respectively. To determine the solubility, the distance of a solvent from the centre of the Hansen solubility sphere,  $R_a$  is calculated by Equation (4). Component A refers to the solute (carotenoids) while component B refers to the solvent. The smaller the  $R_a$ , the greater the affinity between carotenoids and solvent; thus the higher the solubility of carotenoids is in the solvent.<sup>22</sup>

$$R_{a}^{2} = 4 \left( \delta_{d,A} - \delta_{d,B} \right)^{2} + \left( \delta_{p,A} - \delta_{p,B} \right)^{2} + \left( \delta_{h,A} - \delta_{h,B} \right)^{2}$$
(4)

In this work, the four objective functions of the developed solvent are as follow: minimum boiling difference between the solvent and hexane  $(T_{b,diff})$  to ensure the solvent has a convenient  $T_b$  like hexane, minimum heat of vaporization  $(H_v)$  for low energy consumption during solvent recovery process, minimum  $R_a$  for high solubility of carotenoids in the solvent, and minimum total penalty score  $(I_{SHI})$  for an inherently safer and healthier molecule.

Meanwhile,  $F_p$ , octanol-water partition coefficient (log  $K_{ow}$ ), and acute toxicity (96-h  $LC_{50}$  to fathead minnow) are selected as property constraints. Their respective lower bound (LB) and upper bound (UB) are shown in Table 1. GCM models developed by Hukkerikar *et al.*,<sup>16</sup> can be used to estimate  $\delta_d$ ,  $\delta_p$ ,  $\delta_h$ ,  $H_v$  and log  $K_{ow}$ , while the GCM proposed by Martin and Young<sup>23</sup> is utilized to calculate acute toxicity  $LC_{50}$  (96-h to fathead minnow). The selected molecular blocks include CH<sub>3</sub>, CH<sub>2</sub>, CH, C, OH, CH<sub>3</sub>CO, CH<sub>2</sub>CO, CH<sub>3</sub>O, CH<sub>2</sub>O, CHO, CH<sub>3</sub>COO, CH<sub>2</sub>COO, CH<sub>2</sub> (cyclic), CH (cyclic), C (cyclic) and O (cyclic), which are from the molecular fragments of the solvents used for the extraction of carotenoids.

## 3.2 Fuzzy Optimization

The four design objectives are then converted into their respective property operator,  $\Omega_p$  as shown in Table 2.  $\Omega_p$  is equivalent to the simple function f(p) for each target property p, which is exactly the left-hand side of Equation (1). The  $\Omega_p$  range for the four properties to be optimized can be determined by optimizing each of the property one at a time to identify their respective  $V^L$ and  $V^U$ , which are listed in Table 2. In order to minimize the four target properties, the linear membership function of Equation (2) is applied. The optimization model is a mixed integer nonlinear programming (MINLP) as it involves nonlinear constraints when modeling the uncertainty on the allocation of sub-index values to the molecule. Integer cuts are introduced to generate alternate solutions so that there are multiple solvents to conduct experimental verification for their performance.

Table 1: Property constraints for the developed solvent

|                | P     |      |  |
|----------------|-------|------|--|
| Property       | LB    | UB   |  |
| $F_p$ (°C)     | - 1.2 | -    |  |
| log Kow        | -     | 2.86 |  |
| $\log LC_{50}$ | 1.37  | -    |  |
|                |       |      |  |

| Table 2: Target | property operators | for the develo | ped solvent |
|-----------------|--------------------|----------------|-------------|
| $\mathcal{U}$   |                    |                | 1           |

| Property      | $\Omega_p$          | $V^{ m L}$ | $V^{ m U}$ |  |
|---------------|---------------------|------------|------------|--|
| $T_{b, diff}$ | $\exp(T_b/T_{b0})$  | 0.0032     | 2.7261     |  |
| $H_{v}$       | $H_{v} - H_{v_{0}}$ | 14.0244    | 30.8018    |  |
| $R_a$         | $R_a$               | 3.5863     | 9.7779     |  |
| Ishi          | <b>I</b> SHI        | 8.7674     | 11.8096    |  |

#### INTEGRATING SAFETY AND HEALTH INDEXES IN COMPUTER-AIDED MOLECULAR DESIGN UNDER PROPERTY PREDICTION UNCERTAINTY

|         | $\mathcal{O}$ |       |       | 1 1              |            |                           |         |                          |
|---------|---------------|-------|-------|------------------|------------|---------------------------|---------|--------------------------|
| Solvent | λ             | Ishi  | Ra    | $H_{v}$ (kJ/mol) | $T_b$ (°C) | $F_p(^{\circ}\mathrm{C})$ | log Kow | $LC_{50} \text{ (mg/l)}$ |
| A1      | 0.404         | 10.58 | 6.634 | 35.5             | 107.2      | 16.7                      | 1.56    | 68.6                     |
| A2      | 0.379         | 10.30 | 7.432 | 34.3             | 86.2       | 9.6                       | 1.01    | 114.6                    |
| A3      | 0.363         | 10.34 | 7.532 | 35.2             | 101.5      | 16.0                      | 1.20    | 86.3                     |
| A4      | 0.326         | 10.24 | 7.761 | 32.9             | 68.7       | 4.5                       | 0.76    | 208.0                    |
| A5      | 0.215         | 11.15 | 6.271 | 30.8             | 79.7       | 2.1                       | 2.06    | 224.9                    |
| A6      | 0.204         | 11.19 | 6.327 | 31.7             | 95.4       | 8.5                       | 2.25    | 169.3                    |

Table 3: The six generated solvents with their properties

## 4.0 RESULTS AND DISCUSSIONS

The estimated properties of the six synthesized solvents are shown in Table 3, while their molecular structures are illustrated in Figure 4. Solvent A5 has the lowest  $R_a$  and  $H_v$ , while solvent A4 has the lowest  $I_{SHI}$  and  $T_b$  but with the largest  $R_a$ . According to Yara-Varón et al.,<sup>21</sup> they found that solvent A4 (ethyl acetate) offers similar carotenoids extraction yield as compared to that of hexane. Therefore, the other five solvents with lower  $R_a$  than solvent A4 should also exhibit compatible or better extraction yield as compared to hexane. If uncertainty were not managed on the sub-indexes, the original sub-index scores without uncertainty would be applied and solvents A1 to A4 would have a similar ISHI value of 10 while solvents A5 and A6 would have an Ishi score of 11. Based on their scores, the inherent hazard level of solvents A1 to A4 cannot be differentiated, while the same also applies to solvents A5 and A6. Hence, by managing uncertainty on the subindexes, it helps to calculate *I*<sub>SHI</sub> that better represents the actual inherent safety and health hazard level of the molecule.



Figure 4: The generated solvents with their molecular structures

# 5.0 CONCLUSIONS

A novel CAMD method integrating the inherent safety and health sub-indexes has been developed to design molecules with reduced safety and health risks that also achieve a set of desired design properties. Several chemical-related safety and health parameters are adopted from the inherent safety and health indexes as assessment tool to measure the molecular performance. The safety and health attributes of the molecules are measured by their properties, which are estimated through property prediction methods. The uncertainties arised from these prediction models are managed on the sub-indexes to ensure that sub-index values with higher accuracy are allocated to the molecule. This ensures better representation of the safety and health performance for the molecules. A case study on the solvent design for carotenoids extraction from PPF is carried out and fuzzy optimization is applied to develop molecules that simultaneously achieve high functionality and high safety and health performance. The results show that all generated solvents demonstrate favorable performance for the extraction of carotenoids.

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