



# **Development and Characterization of Co-Cr-Mo (F-75 alloy)/hydroxyapatite Composites Fabricated by Powder Metallurgy for Biomedical Applications**

By

**Nur Maizatul Shima Bt. Adzali  
(0840410226)**

A thesis submitted in fulfilment of the requirements for the degree of  
Doctor of Philosophy

**School of Materials Engineering  
UNIVERSITI MALAYSIA PERLIS**

2013

## **ACKNOWLEDGEMENT**

In the name of Allah S.W.T and praise to Allah as our God, the Most Gracious and the Most Merciful.

Alhamdulilah, all praises to Allah for the strengths and His blessing in completing this thesis. Above all, I would like to thank Prof. Dr. Shamsul Baharin Jamaludin, my principal supervisor, for his guidance, help and support throughout this research. This thesis would not have been possible without his help. Special appreciation also goes to my co-supervisor, Dr. Mohd Nazree Derman for his guidance and knowledge regarding this topic.

I would like to acknowledge the University Malaysia Perlis (UniMAP), especially Schools of Materials Engineering for giving me the opportunity to complete my research. This gratitude is also goes to Dean Schools of Materials Engineering, Dr. Khairel Rafezi Ahmad and all the staffs, especially Dr. Banjuraizah, Dr. Sobri, Mr. Saiful Rizam, Mr. Mohd Nasir, Mr. Ahmad Hadzrul Iqwan, Mr. Norzaidi, Mr. Ku Hasrin, Mr. Azmi, and Mr. Faisal. Other than that, I am also thankful to the dedicated staff from other schools that helped me a lot during my lab work, which were Mr. Mohd Affandi Derman, from Schools of Environmental Engineering, and Mrs. Mira, from Institute of Nano Electronic Engineering (INEE), UniMAP. A special thanks also to UniMAP library for superb facilities.

I am extremely grateful to my friends, especially Dr. Zuraidawani, Mrs. Nurhidayah, Miss Rohaya, Miss Norazimah, Mr. Chan, Miss Zatil, Mrs. Norwanis and Miss Fariza for their friendship and full support throughout my journey to complete my study. Thank you for always staying by my side during these tough years.

Finally, sincere thank and appreciation goes to my parents Mr. Adzali Khalit and Mrs. Che Maznah Hj. Ashaari, for their prayers, unconditional support, understanding and encouragement. I wish to express very special thank to my lovely husband, Mr. Azrul Sani Ismail and also for my two sons, Aiman Husaini and Ieman Abqary. Thank you very much for being my other half during the wonderful times that we have passed together. To those who indirectly contributed in this research, your kindness means a lot to me.

Thank you very much.

NUR MAIZATUL SHIMA ADZALI

2013.

## TABLES OF CONTENTS

	PAGE
<b>THESIS DECLARATION</b>	i
<b>ACKNOWLEDGEMENT</b>	ii
<b>TABLE OF CONTENTS</b>	iv
<b>LIST OF FIGURES</b>	x
<b>LIST OF TABLES</b>	xvii
<b>LIST OF ABBREVIATIONS</b>	xix
<b>LIST OF SYMBOLS</b>	xxii
<b>ABSTRAK</b>	xxiii
<b>ABSTRACT</b>	xxiv
<b>CHAPTER 1: INTRODUCTION</b>	
1.1    Biomaterials	1
1.2    Problem statements	5
1.3    Objectives of the Study	7
1.4    Scope of Work	8
<b>CHAPTER 2: LITERATURE REVIEW</b>	
2.1    A Brief Highlight of Biomaterials Development	9

2.2	Requirements of Biomaterials	10
2.2.1	Mechanical properties requirements	11
2.2.1.1	Adequate strength	11
2.2.1.2	Modulus equivalent to bone	12
2.2.2	Non mechanical requirements	13
2.2.2.1	High corrosion resistance	13
2.2.2.2	High wear resistance	14
2.2.2.3	Biocompatibility	14
2.2.3	Osseointegration	15
2.3	Group of Biomaterials	15
2.3.1	Metallic Biomaterials	18
2.3.1.1	Co-Cr based alloys	20
2.3.1.2	Titanium alloys	23
2.3.1.3	Magnesium alloys	24
2.3.1.4	Stainless steels	24
2.3.1.5	Promising materials	25
2.3.2	Ceramic	25
2.3.2.1	Hydroxyapatite	26
2.3.2.2	Bioactive glass and glass ceramics	31
2.3.3	Polymer	31
2.3.4	Composite	32
2.3.4.1	Co based alloy composite	33
2.3.4.2	Ti based alloy composite	35
2.3.4.3	Mg based alloy composite	36
2.4	Powder Metallurgy	37

2.4.1	Powder metallurgy for biomaterials	38
2.4.2	Basic steps of powder metallurgy process	40
2.4.2.1	Manufacturing of metal powders	40
2.4.2.2	Blending and mixing	41
2.4.2.3	Compacting	41
2.4.2.4	Sintering	42
2.5	Corrosion Studies	44
2.5.1	Corrosion of bimetallic materials	44
2.5.2	Electrochemical reactions	46
2.5.3	Corrosion rate	48
2.6	Bioactivity Test	51
2.6.1	Introduction to bioactivity test	51
2.6.2	Bioactive materials in vitro	52

### **CHAPTER 3 : RESEARCH METHODOLOGY**

3.1	Background of the Research	55
3.2	Introduction	57
3.3	Raw Materials	58
3.3.1	Co-Cr-Mo alloy (F-75)	59
3.3.2	Hydroxyapatite (HAP)	60
3.3.3	Stearic acid	60
3.4	Raw Materials Characterization	61
3.4.1	Morphological study by SEM	61
3.4.2	X-Ray diffraction (XRD)	61
3.4.3	Particle size analysis	62

3.5. Fabrication of the F-75/HAP Composites	62
3.5.1 Weighing and milling	62
3.5.2 Compacting	63
3.5.2.1 Set up of tool steel die	64
3.5.3 Sintering	65
3.5.3.1 Set up of tube furnace	65
3.5.3.2 Sintering process	66
3.6 Physical and Mechanical Properties Testing	67
3.6.1 Theoretical density, bulk density and apparent porosity	67
3.6.2 Compression test	69
3.6.2.1 Fracture mode after compression test	69
3.7 Microstructural Observation	69
3.7.1 Sample preparation for metallographic study	70
3.7.2 Optical microscope observation	70
3.7.3 SEM observation	71
3.7.4 SEM-EDS line scanning	71
3.8 Corrosion Test	72
3.8.1 Sample preparation for electrochemical test	72
3.8.2 Potentiodynamic polarization	73
3.8.3 Corrosion rate	75
3.9 Bioactivity Test	77
3.9.1 The procedure of bioactivity test	77
3.9.2 pH measurement	79
3.9.3 Fourier Transform Infra-red analysis (FTIR)	80

## **CHAPTER 4: RESULTS AND DISCUSSION**

4.1	Raw Materials Characterization	81
4.1.1	Co-Cr-Mo (F-75) alloys	81
4.1.1.1	Morphology study by SEM for initial Co-Cr-Mo powder	81
4.1.1.2	Particle size analysis for initial Co-Cr-Mo powder	82
4.1.2	Hydroxyapatite (HAP)	83
4.1.2.1	Morphology study by SEM for initial HAP powder	83
4.1.2.2	Particle size analysis for initial HAP powder	84
4.2	Physical and Mechanical Properties of the Composites	85
4.2.1	Bulk density and apparent porosity	85
4.2.2	Compressive strength	90
4.2.2.1	Fractography after compression test	92
4.3	Microstructural Characterization	98
4.3.1	Optical microscope observation	98
4.3.2	SEM observation	102
4.3.3	SEM-EDS for line scan analysis	107
4.4	X-ray Diffraction analysis for F-75/HAP composite after sintering	112
4.5	Corrosion Test	117
4.5.1	Electrochemical test	117
4.5.1.1	Tafel plot	121

4.5.1.2 Microstructural observation after electrochemical	
test	125
4.6     Bioactivity Test	132
4.6.1 XRD analysis	132
4.6.2 SEM analysis	138
4.6.3 FTIR analysis	147
4.6.4 pH analysis	154
<b>CHAPTER 5: CONCLUSIONS</b>	
5.1 Conclusions	159
5.2 Future Work	161
<b>REFERENCES</b>	162
<b>APPENDICES</b>	174
<b>GLOSSARY</b>	177
<b>LIST OF PUBLICATIONS</b>	179

## LIST OF FIGURES

NO.		PAGE
2.1	Total knee and hip implants components (Nasab & Hassan, 2010)	13
2.2	Clinical uses of inorganic biomaterials (Hench, 1985)	16
2.3	Crystal structure of hydroxyapatite (Vallet-Regi & Gonzalez-Calbet, 2004)	28
2.4	Compressive stress-strain curves for AZ91 (magnesium) and AZ91-20FA (magnesium reinforced with 20 wt. % FA) samples (Razavi et al., 2010)	37
2.5	Basic steps in making parts by powder metallurgy (Budinski and Budinski, 2002)	40
2.6	Steps in pressing operations using tool steel die (Rao, 2002)	42
2.7	Two spheres point contact sintering model (Tang et al., 2002)	43
2.8	Factors and their effects on biocompatibility (Singh & Dahotre, 2007)	45
2.9	Tafel plot for sample AZ91 (magnesium alloys) and AZ91-20FA (Magnesium alloys with 20 wt. % FA) (Razavi et al., 2010)	46
2.10	Corrosion process showing anodic and cathodic currents components (Willert-Porada, 2011)	47
2.11	SEM microphotographs of all composite materials before and after immersion in SBF for various periods of time (Roumeli et al., 2011)	53
3.1	Phases in the lab work for this research	58
3.2	Schematic diagram of the die	63

3.3	Sintering equipment was set up, with tube furnace used to sinter the samples	65
3.4	Sintering profile for the sintering process at 1100 <sup>0</sup> C	66
3.5	Sample preparation for electrochemical test	73
3.6	Schematic diagram of potentiodynamic polarization test in 0.9 wt. % NaCl	74
3.7	Theoretical Tafel plots illustrating the Tafel polarization method (Mansfeld, 2002)	75
3.8	Sample immersed in the SBF solution at 36.5 <sup>0</sup> C in a water bath	79
4.1	Co-Cr-Mo alloy powders in 500X magnification	82
4.2	The result for particle size distribution of Co-Cr-Mo alloy powder	83
4.3	The micrograph of initial hydroxyapatite powder in 500X magnification	84
4.4	Particle size distribution of hydroxyapatite powder	85
4.5	The value of theoretical density and bulk density of the F-75/HAP composites after sintered at three different temperature and six different wt. % of HAP	87
4.6	The value of apparent porosity of the F-75/HAP composites after sintered at three different temperature and six different wt. % of HAP	88
4.7	The plot of compressive strength vs different addition of HAP (in wt. %) for three different sintering temperature	91
4.8	SEM micrographs of the samples sintered at 1100 <sup>0</sup> C after compression test with (a) no HAP (b) 2 wt. % HAP (c) 4 wt. % HAP (d) 6 wt. % HAP (e) 8 wt. % HAP f) 10 wt. % HAP	95

4.9	SEM micrographs of the samples sintered at 1150 <sup>0</sup> C after compression test with (a) no HAP (b) 2 wt. % HAP (c) 4 wt. % HAP (d) 6 wt. % HAP (e) 8 wt. % HAP (f) 10 wt. % HAP	96
4.10	SEM micrographs of the samples sintered at 1200 <sup>0</sup> C after compression test with (a) no HAP (b) 2 wt. % HAP (c) 4 wt. % HAP (d) 6 wt. % HAP (e) 8 wt. % HAP (f) 10 wt. % HAP	97
4.11	Optical micrographs of F-75/HAP composites sintered at 1100 <sup>0</sup> C for six different addition of HAP, (a) no HAP (b) 2 wt. % HAP (c) 4 wt. % HAP (d) 6 wt. % HAP (e) 8 wt. % HAP (f) 10 wt. % HAP	99
4.12	Optical micrographs of F-75/HAP composites sintered at 1150 <sup>0</sup> C for six different addition of HAP, (a) no HAP (b) 2 wt. % HAP (c) 4 wt. % HAP (d) 6 wt. % HAP (e) 8 wt. % HAP (f) 10 wt. % HAP	100
4.13	Optical micrographs of F-75/HAP composites sintered at 1200 <sup>0</sup> C for six different addition of HAP, (a) no HAP (b) 2 wt. % HAP (c) 4 wt. % HAP (d) 6 wt. % HAP (e) 8 wt. % HAP (f) 10 wt. % HAP	101
4.14	SEM micrograph for sample with (a) no HAP, (b) 2 wt. % HAP (c) 4 wt. % HAP, (d) 6 wt. % HAP, (e) 8 wt. % HAP, (f) 10 wt. % HAP after sintered at 1100 <sup>0</sup> C.	103
4.15	SEM micrograph for sample with (a) no HAP, (b) 2 wt. % HAP (c) 4 wt. % HAP, (d) 6 wt. % HAP, (e) 8 wt. % HAP (f) 10 wt. % HAP after sintered at 1150 <sup>0</sup> C.	104

4.16	SEM micrograph for sample with (a) no HAP, (b) 2 wt. % HAP (c) 4 wt. % HAP, (d) 6 wt. % HAP, (e) 8 wt.% HAP (f) 10 wt.% HAP after sintered at 1200 <sup>0</sup> C	105
4.17	SEM image and EDS line scan result of sample F-75/6HAP after sintered at 1100 <sup>0</sup> C	109
4.18	SEM image and EDS line scan result of sample F-75/10HAP after sintered at 1100 <sup>0</sup> C	109
4.19	SEM image and EDS line scan result of sample F-75/6HAP after sintered at 1150 <sup>0</sup> C	110
4.20	SEM image and EDS line scan result of sample F-75/10HAP after sintered at 1150 <sup>0</sup> C	110
4.21	SEM image and EDS line scan result of sample F-75/6HAP after sintered at 1200 <sup>0</sup> C	111
4.22	SEM image and EDS line scan result of sample F-75/10HAP after sintered at 1200 <sup>0</sup> C	111
4.23	XRD pattern for F-75/HAP composite after sintered at temperature 1100 <sup>0</sup> C	113
4.24	XRD pattern for F-75/HAP composite after sintered at temperature 1150 <sup>0</sup> C	114
4.25	XRD pattern for F-75/HAP composite after sintered at temperature 1200 <sup>0</sup> C	115
4.26	XRD patterns for raw material of (a) F-75 and (b) HAP before sintering	116
4.27	Tafel plots for samples sintered at 1100 <sup>0</sup> C in 0.9 wt. % of NaCl	123
4.28	Tafel plots for samples sintered at 1150 <sup>0</sup> C in 0.9 wt. % of NaCl	124

4.29	Tafel plots for samples sintered at 1200 <sup>0</sup> C in 0.9 wt. % of NaCl	125
4.30	SEM photomicrographs of sample with (a) no HAP (b) 2 wt. % HAP (c) 4 wt. % HAP (d) 6 wt. % HAP (e) 8 wt. % HAP (f) 10 wt. % HAP after the potentiodynamic examinations after sintered at 1100 <sup>0</sup> C	127
4.31	SEM photomicrographs of sample with (a) no HAP (b) 2 wt. % HAP (c) 4 wt. % HAP (d) 6 wt. % HAP (e) 8 wt. % HAP (f) 10 wt. % HAP after the potentiodynamic examinations after sintered at 1150 <sup>0</sup> C	128
4.32	SEM photomicrographs of sample with (a) no HAP (b) 2 wt. % HAP (c) 4 wt. % HAP (d) 6 wt. % HAP (e) 8 wt. % HAP (f) 10 wt. % HAP after the potentiodynamic examinations after sintered at 1200 <sup>0</sup> C	129
4.33	EDS analysis for (a) F-75/0HAP (b) F-75/8 HAP sintered at 1100 <sup>0</sup> C, after electrochemical test in 0.9 wt. % NaCl	130
4.34	EDS analysis for (a) F-75/0HAP (b) F-75/8 HAP sintered at 1150 <sup>0</sup> C, after electrochemical test in 0.9 wt. % NaCl	130
4.35	EDS analysis for (a) F-75/0HAP (b) F-75/8 HAP sintered at 1200 <sup>0</sup> C, after electrochemical test in 0.9 wt. % NaCl	131
4.36	XRD pattern for F-75/HAP composite sintered at 1100 <sup>0</sup> C after immersion in SBF for 18 days	134
4.37	XRD pattern for F-75/HAP composite sintered at 1150 <sup>0</sup> C after immersion in SBF for 18 days	135
4.38	XRD pattern for F-75/HAP composite sintered at 1200 <sup>0</sup> C after immersion in SBF for 18 days	136

4.39	SEM microphotographs and EDS analysis of samples F-75/HAP sintered $1100^0\text{C}$ after immersion in SBF for 18 days with (a) no HAP (b) 2 wt. % HAP (c) 4 wt. % HAP (d) 6 wt. % HAP (e) 8 wt.% HAP (f) 10 wt.% HAP	141
4.40	SEM microphotographs and EDS analysis of samples F-75/HAP sintered at $1150^0\text{C}$ after immersion in SBF for 18 days with (a) no HAP (b) 2 wt. % HAP (c) 4 wt. % HAP (d) 6 wt. % HAP (e) 8 wt.% HAP (f) 10 wt.% HAP	143
4.41	SEM microphotographs and EDS analysis of samples F-75/HAP sintered at $1200^0\text{C}$ after immersion in SBF for 18 days with (a) no HAP (b) 2 wt. % HAP (c) 4 wt. % HAP (d) 6 wt. % HAP (e) 8 wt.% HAP (f) 10 wt.% HAP	145
4.42	FTIR spectrums of SBF solution for F-75/0HAP sintered at $1100^0\text{C}$ before and after immersion in SBF for various periods of time	149
4.43	FTIR spectrums of SBF solution for F-75/4HAP sintered at $1100^0\text{C}$ before and after immersion in SBF for various periods of time	149
4.44	FTIR spectrums of SBF solution for F-75/8HAP sintered at $1100^0\text{C}$ before and after immersion in SBF for various periods of time	150
4.45	FTIR spectrums of SBF solution for F-75/0HAP sintered at $1150^0\text{C}$ before and after immersion in SBF for various periods of time	150
4.46	FTIR spectrums of SBF solution for F-75/4HAP sintered at $1150^0\text{C}$ before and after immersion in SBF for various periods of time	151
4.47	FTIR spectrums of SBF solution for F-75/8HAP sintered at $1150^0\text{C}$ before and after immersion in SBF for various periods of time	151
4.48	FTIR spectrums of SBF solution for F-75/0HAP sintered at $1200^0\text{C}$	

before and after immersion in SBF for various periods of time	152
4.49 FTIR spectrums of SBF solution for F-75/4HAP sintered at 1200 <sup>0</sup> C before and after immersion in SBF for various periods of time	152
4.50 FTIR spectrums of SBF solution for F-75/8HAP sintered at 1200 <sup>0</sup> C before and after immersion in SBF for various periods of time	153
4.51 Change of pH of the SBF with time after the immersion of samples treated F-75/0HAP, F-75/2HAP, F-75/4HAP, F-75/6HAP, F-75/8HAP, and F-75/10HAP for sample sintered at 1100 <sup>0</sup> C	156
4.52 Change of pH of the SBF with time after the immersion of samples treated F-75/0HAP, F-75/2HAP, F-75/4HAP, F-75/6HAP, F-75/8HAP, and F-75/10HAP for sample sintered at 1150 <sup>0</sup> C	156
4.53 Change of pH of the SBF with time after the immersion of samples treated F-75/0HAP, F-75/2HAP, F-75/4HAP, F-75/6HAP, F-75/8HAP, and F-75/10HAP for sample sintered at 1200 <sup>0</sup> C	157

## LIST OF TABLES

NO.	PAGE
2.1 Properties of bone (Yamada, 1970)	12
2.2 Materials used in implants (Bhat, 2005)	16
2.3 Summary of different metallic materials used or developed for orthopedic applications, including their elastic modulus, yield strength and ultimate strength (Navarro et al., 2008)	19
2.4 Mechanical properties requirements for Co-Cr based alloys (Wong & Bronzino, 2007)	21
2.5 Mechanical properties of Co-Cr based alloys and human bone (Mudali et al., 2003)	21
2.6 Family of calcium phosphate compounds (Shi, 2004)	27
2.7 Composition of biological apatites and hydroxyapatite (Wong & Bronzino, 2007)	29
2.8 Mechanical Properties of Hydroxyapatite (Hench, 1971)	29
2.9 The comparison of elastic modulus values of bone and investigated materials (GPa) (Gradzka et al., 2008)	34
2.10 Values of the constants used in the Faraday Equation Rate (ASTM G 102-89)	50
2.11 Ion concentrations of SBF and human blood plasma (BS ISO 23317: 2007)	51
3.1 The characteristics and chemical composition of Co-Cr-Mo (F-75) alloy powder (Sandvik Osprey Ltd., UK.)	59
3.2 Properties of Co-Cr-Mo alloy (Wong & Bronzino, 2007)	59

3.3	Physical Properties of Hydroxyapatite Powder (Yoshimura & Suda, 1994)	60
3.4	The composition of the composites	62
4.1	Electrochemical parameter of samples after sintered at 1100 <sup>0</sup> C for F-75/HAP composites after electrochemical test in 0.9 wt. % NaCl	119
4.2	Electrochemical parameter of samples after sintered at 1150 <sup>0</sup> C for F-75/HAP composites after electrochemical test in 0.9 wt. % NaCl	119
4.3	Electrochemical parameter of samples after sintered at 1200 <sup>0</sup> C for F-75/HAP composites after electrochemical test in 0.9 wt. % NaCl	120

## LIST OF ABBREVIATIONS

A/W	Apatite/Wollastonite
AISI 316L	0.03 wt. % C, 17-20 wt. % Cr, 12-14 wt. % Ni, 2-3 wt. % Mo and minor amounts of nitrogen, manganese, phosphorus, silicon and sulphur
ASTM	American Society for Testing and Materials
AZ91-HAP	Magnesium-Hydroxyapatite
B <sub>4</sub> C	Boron carbide
C	Carbon
Ca <sub>2</sub> P <sub>2</sub> O <sub>7</sub>	Calcium Pyrophosphate
CaCO <sub>3</sub>	Calcium Carbonate
CaHPO <sub>4</sub> .2H <sub>2</sub> O	Calcium Hydrogen Phosphate
Ca <sub>10</sub> (PO <sub>4</sub> ) <sub>6</sub> (OH) <sub>2</sub>	Hydroxyapatite
CeO <sub>2</sub>	Ceria
CMC	Ceramic matrix composites
Co	Cobalt
CO <sub>2</sub>	Carbon dioxide
Co-Cr	Cobalt-Chromium
Co-Cr-Mo	Cobalt-Chromium-Molybdenum
Cr	Chromium
CR	Corrosion Rate
E <sub>corr</sub>	Corrosion current
EW	Equivalent weight

FA	Fluoroapatite
FCC	Face centre cubic
Fe	Ferum
H <sub>2</sub> O	Water
HAP	Hydroxyapatite
HCAp	Hydroxyl carbonate apatite
HCP	Hexagonal close-packed
Mg	Magnesium
MMC	Metal Matrix Composites
Mn	Manganese
Mo	Molybdenum
Ni	Nickel
Ni Ti	Nickel-titanium
OH <sup>-</sup>	Hydrogen ion
PE	Polyethylene
PGA	Polyglycolide
PLA	Polylactic acid
PM	Powder Metallurgy
PMC	Polymer matrix composite
PMMA	Polymethyl-methacrylate
PP	Polypropylene
Pt	Platinum
R <sub>p</sub>	Polarization resistance ( $\Omega \cdot \text{cm}^2$ ).
rpm	Rotation Per Minute
SBF	Simulated Body Fluid

Si	Silicon
SiC	Silicon carbide
$\text{Si}_3\text{N}_4$	Silicon Nitride
SMA	Shape Memory Alloy
TAP	Transferred Arc Plasma
Ti	Titanium
Ti-6Al-4V-HAP	Titanium-6Aluminum-4Vanadium-Hydroxyapatite
Ti-HAP	Titanium-Hydroxyapatite

## LIST OF SYMBOLS

%	Percent
$\Omega \cdot \text{cm}^2$	Ohm centimetre square
${}^\circ\text{C}$	Degree Celsius
A	exposed specimen area, $\text{cm}^2$
$\text{g}/\text{cm}^3$	gram per centimetre cube
$\text{GN}/\text{m}^2$	Giga Newton per metre square
GPa	Giga Pascal
KN	Kilo Newton
min	minutes
MPa	Mega Pascal
mpy	Mil per year
$N_0$	Rotational speed
wt. %	Weight Percent
$\mu\text{m}$	micrometer

## **Pembangunan dan Pencirian Komposit Co-Cr-Mo (Aloi F-75)/Hidroksiapatit yang Difabrikasi Menggunakan Metalurgi Serbuk Untuk Aplikasi Bioperubatan**

### **ABSTRAK**

Aloi Co-Cr-Mo (F-75) telah terkenal digunakan dalam bidang bioperubatan kerana biokeserasian yang amat baik apabila digunakan di dalam badan manusia atau haiwan. Serbuk hidroksiapatit (HAP) telah digunakan sebagai bahan pengisi kerana HAP adalah salah satu bahan yang mempunyai biokeserasian yang amat efektif bersamaan dengan kandungan mineral untuk tulang dan gigi. Kajian ini melaporkan tentang fabrikasi dan pencirian aloi F-75 ditambah dengan HAP yang disediakan menggunakan kaedah metalurgi serbuk. Kajian ini dijalankan dengan tumpuan kepada kesan penambahan HAP kepada aloi F-75 dan suhu persinteran terhadap sifat-sifat fizikal dan mekanikal komposit F-75/HAP, mikrostruktur, serta kelakuan kakisan dan bio-aktiviti komposit ini. Dalam fabrikasi komposit F-75/HAP, 2, 4, 6, 8 dan 10 % berat HAP telah ditambah kepada aloi F-75. Sampel rujukan aloi F-75 (tanpa penambahan HAP) juga telah disediakan untuk semua suhu persinteran. Campuran diadun menggunakan mesin putaran selama 20 minit pada 154 rpm sebelum ditekan sejuk pada 550 MPa menggunakan mesin penekanan searah. Sampel disinter pada tiga suhu persinteran yang berbeza ( $1100^{\circ}\text{C}$ ,  $1150^{\circ}\text{C}$  dan  $1200^{\circ}\text{C}$ ) di dalam relau tiub selama 2 jam. Sifat-sifat fizikal ditentukan melalui ujian ketumpatan pukal dan keliangan ketara, manakala sifat mekanikal ditentukan melalui ujian kekuatan mampatan. Kelakuan kakisan komposit F-75/HAP dianalisis menggunakan ujian elektrokimia yang dikawal oleh potensiostat Gamry G300. Kelakuan bio-aktiviti komposit ini telah dijalankan secara in-vitro dengan merendam komposit ke dalam bendalir badan tersimulasi selama 18 hari. Analisis XRD, SEM, FTIR dan pH dilakukan untuk menentukan kehadiran lapisan apatit di permukaan komposit F-75/HAP. Daripada kajian ini, nilai ketumpatan pukal berkurang apabila kandungan HAP bertambah. Nilai tertinggi ketumpatan pukal adalah pada komposit yang mengandungi 2% berat HAP dengan ketumpatan  $6.6217 \text{ g/cm}^3$  dengan suhu persinteran  $1200^{\circ}\text{C}$ , manakala nilai ketumpatan pukal terendah adalah pada komposit dengan penambahan 10% berat HAP selepas disinter pada suhu  $1150^{\circ}\text{C}$  ( $4.3915 \text{ g/cm}^3$ ). Keliangan ketara sampel yang disinter menunjukkan bahawa apabila penambahan HAP ditingkatkan, keliangan ketara akan meningkat di dalam julat 13.13% (untuk 2% berat HAP) ke 37.58% (untuk 10 % berat HAP). Kekuatan mampatan berkurang dengan penambahan HAP. Sampel dengan penambahan 2 % berat HAP dengan suhu persinteran  $1200^{\circ}\text{C}$ , memberi nilai kekuatan mampatan yang paling tinggi (341.81 MPa). Mikrostruktur komposit F-75/HAP selepas disinter pada tiga suhu persinteran yang berbeza menunjukkan keliangan dan kegumpalan HAP meningkat mengikut suhu persinteran. Keputusan ujian kakisan menunjukkan sampel dengan penambahan 8 % berat HAP memberikan nilai kadar kakisan yang paling rendah ( $16.59 \times 10^{-6} \text{ mpy}$  untuk F-75/8%HAP yang disinter pada  $1150^{\circ}\text{C}$ ). Daripada keputusan ujian bio-aktiviti, lapisan apatit karbonat telah terbentuk pada permukaan komposit. Berdasarkan keputusan ujian sifat fizikal dan mekanikal komposit, penambahan HAP yang optimum terhadap aloi F-75 adalah 2 % berat, manakala sampel yang disinter pada suhu tinggi ( $1200^{\circ}\text{C}$ ) menunjukkan sifat fizikal, mekanikal dan juga kelakuan kakisan yang baik. Daripada ujian kakisan, komposit F-75/6%HAP dan F-75/8%HAP yang disinter pada suhu tinggi menghasilkan kerintangan kakisan yang baik. Aloi F-75 yang biolengai boleh ditukarkan kepada bioaktif dengan menambahkan sehingga 10 % berat HAP.

# **Development and Characterization of Co-Cr-Mo (F-75 alloy)/hydroxyapatite Composites Fabricated by Powder Metallurgy for Biomedical Applications**

## **ABSTRACT**

Co-Cr-Mo (F-75) alloy is known to be used in biomedical field because of their excellent biocompatibility when implanted to human or animal body. Hydroxyapatite (HAP) powders have been used as filler because HAP is the one of the most effective biocompatible materials with similarities to mineral constituents of bones and teeth. This research reported the fabrication and characterization of F-75 alloy filled with HAP which have been prepared by powder metallurgy method. This study has focused on the effect of HAP addition into F-75 alloy and sintering temperature on the physical and mechanical properties of the F-75/HAP composites, its microstructure, and also its corrosion and bioactivity behaviour. In fabrication of the F-75/HAP composite, 2, 4, 6, 8 and 10 wt. % of HAP have been added to F-75 alloys. The reference samples of F-75 alloy (with no addition of HAP) also have been prepared for all sintering temperatures. The mixtures were milled on a rotation mill for 20 minutes at 154 rpm before cold compacted at 550 MPa using an uniaxial press machine. The samples then have been sintered at three different sintering temperatures ( $1100^{\circ}\text{C}$ ,  $1150^{\circ}\text{C}$  and  $1200^{\circ}\text{C}$ ) in a tube furnace for 2 hours. Physical properties were measured by means of bulk density and apparent porosity while mechanical property was measured in term of compressive strength. The corrosion behaviour of the F-75/HAP composite has been analysed using electrochemical test controlled by Gamry G300 potentiostat. Bioactivity test for the composite was conducted in-vitro by immersing the composite into simulated body fluid for 18 days. XRD, SEM, FTIR and pH analyses had been done in order to observe the presence of the apatite layer on the surface of F-75/HAP composites. From this study, the values of bulk density decreased as the HAP content increased. The highest value of bulk density was gained by the composite with 2 wt. % of HAP with value  $6.6217 \text{ g/cm}^3$  with sintering temperature  $1200^{\circ}\text{C}$ , while the lowest bulk density value was given by the composite with 10 wt. % of HAP after sintered at  $1150^{\circ}\text{C}$  ( $4.3915 \text{ g/cm}^3$ ). The apparent porosity was increased in the range of 13.13% (for 2 wt. % HAP) to 37.58% (for 10 wt. % HAP). Compressive strength was decreased by the additional of HAP. The sample with 2 wt. % of HAP addition with sintering temperature  $1200^{\circ}\text{C}$  gave the highest compressive strength (341.81 MPa). The microstructure of F-75/HAP composites after sintering at three different sintering temperatures showed that porosity and HAP agglomeration increased with HAP content and sintering temperature. The results of corrosion test showed that the samples with 8 wt. % HAP addition gave the lowest value for corrosion rate ( $16.59 \times 10^{-6} \text{ mpy}$  for F-75/8% HAP sintered at  $1150^{\circ}\text{C}$ ). From bioactivity test results, the carbonated apatite layer was formed on the surfaces of the composite. According to the results for physical and mechanical properties testing of the composites, the optimum HAP addition to F-75 alloy was 2 wt. %, while samples that have been sintered at higher temperature ( $1200^{\circ}\text{C}$ ), showed good physical and mechanical properties and also corrosion behavior. From corrosion test, F-75/6% HAP and F-75/8% HAP composites that have been sintered at higher temperature showed good corrosion resistance. Bioinert F-75 alloys can be converted into F-75 bioactive type by adding up to 10 wt. % of HAP.