

Universiti Malaysia Perlis

INVENTORS DR. NORSURIA MAHMED PN. SRI ASLIZA MD AMIN

CONTACT DETAILS SCHOOL OF MATERIALS ENGINEERING UNIVERSTIT MALAYSIA PERLIS TAMAN MUHIBBAH, S2000 JEJAWI, PERLIS. s-meilt: norsuurla@winimap.edu.my

MULTIFUNCTIONAL MAGNETIC CORE NANOPARTICLES

PROBLEM STATEMENT

- Needs for a nanoparticle that can be used for multifunctional applications.
- Green technology: needs for functional nanoparticles that can be recycled.

PRODUCT DESCRIPTION

In this product, a magnetic material such as magnetite (Fe_3O_4) nanoparticles has been used as a core material. The functionality of this particles is multiplied by coating/combining with various functional nanoparticles such as silica, silver and silver chloride by a simple wet chemistry process, in the form of coreshell structures. Due to its good, magnetic behavior, magnetite nanoparticles can act as an 'anchor' to direct the movement of the 'attached' functional nanoparticles by applying a magnetic field. Depending on the ratio of the materials used, and the way they are combined, it is possible to create applications for different needs.

NOVELTIES

- The combination of 4 different functional nanoparticles (magnetite, silica, silver, silver chloride) under one particle system.
- Doping of 50 nm silver chloride nanocubes on silica surface.

ADVANTAGES

- Multifunctional magnetite nanoparticles can be recycled: reduced the re-production cost and support the green technology policy.
- The simple, room temperature synthesis is economically feasible: low production cost (time, energy, money).

COLLABORATIONS





PUBLICATIONS

- N. Mahmed, O. Heczko, A. Lancok, S-P. Hannula (2014). The magnetic and oxidation behavior of bare and silica-coated iron oxide nanoparticles synthesized by reverse co-precipitation of ferrous ion (Fe2+) in ambient atmosphere. *Journal* of Magnetism and Magnetic Materials, 353, 15-22.
 N. Mahmed, J. Larismaa, O. Heczko, M. E. Cura, S-P. Hannula (2013). Influence of sintering temperature on the
- N. Mahmed, J. Larismaa, O. Heczko, M. E. Cura, S-P. Hannula (2013). Influence of sintering temperature on the properties of pulsed electric current sintered hybrid coreshell powders. *Journal of the European Ceramic Society*, 33, 2233-2239.
 N. Mahmed, Oleg Heczko, Simo-Pekka Hannula (2013). Influence of hydrochloric acid concentrations on the formation
- N. Mahmed, Oleg Heczko, Simo-Pekka Hannula (2013). Influence of hydrochloric acid concentrations on the formation of AgCi-doped iron oxida-silica coreshell structures. Advances in Science and Technology 77, 184-189
 N. Mahmed, H. Jiang, O. Heczko, O. Söderberg, S.P. Hannula (2012). Influence of different synthesis approach on doping behavior of silver nanoparticles onto the iron-oxide silica coreshell surfaces. Journal of Nanoparticle Research,
- Joping behavior of silver nanoparticles onto the iron-oxide silica coreshell surfaces. Journal of Nanoparticle Research, 10:037 Manual, O. Heczko, R. Maki, O. Söderberg, E. Haimi, S-P. Hannula (2012). Novel iron oxide-silica coreshell powders
- Compared to C. Flex.cov, rt. man, o. -ouer berg, c. namm, -rr. namma (2012). Novel ron oxide-silica coreshell powders compared the using pulsed electric current sintering: Optical and magnetic properties. *Journal of the European* Comm. Science 2:281-2968



PRODUCT

Magnetite nanoparticles



Magnetite-silica coreshell structure



Magnetite-silica-silver



Magnetite-silica-silver chloride-silver



Magnetite-silica-silver chloride-silver

COMMERCIAL POTENTIAL



ERSITY AALTO UNIVERSITY AAL

RSITY AALTO UNIVERSITY AAL

RSITY AALTO UNIVERSITY AAL

RSITY AALTO UNIVERSITY AAL

AALTO UND

INTY AALTO UN

RSTTY ALLTO UNIV

RSITE

Magnetic nanoparticle attached with drugs/medicine

Radioactive

ALAYSIA PERLIS



2. Magneto-optical media

AL

M



3. Nanoprobe for biomedical application, nanobiosensor, antimicrobial activity

4. Photocatalyst, Antimicrobial activity

- Promote slow release of Ag+ for antimicrobial property
- As photocatalyst material absorb in UV light
- Combine with silver: plasmonic photocatalyst – absorb in UV-Visible light



Reaction with magnet

NI@SH CO C

Knowlette Variation - Excellence